

LEVELS OF LEAD, COPPER AND IRON IN FISH FROM SOME STREAMS WITHIN A RAVINE IN UYO MUNICIPALITY, NIGERIA.

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ABSTRACT

The levels of three toxic metals (Pb, Cu and Fe) were determined in seven species of fish: *Malapterurus electricus* (F₁), *Auchenoglanis fasciatus* (F₂), *Brienomyrus brachyistius* (F₃), *Chromidotilapia guntheri* (F₄), *Hopsetus* (F₅), *Papyrocranus afer* (F₆), and *Bryarius longipinnis* (F₇) from some streams within a ravine in Uyo municipality in Akwa Ibom State of Nigeria. The results obtained showed very high mean concentrations of Pb (13.12 $\mu\text{g g}^{-1}$) in F₁, (12.04 $\mu\text{g g}^{-1}$) in F₆ and (8.76 $\mu\text{g g}^{-1}$) in F₇ samples while levels of Fe were high only in F₁ (11.03 $\mu\text{g g}^{-1}$), F₅ (11.23 $\mu\text{g g}^{-1}$) and F₇ (9.71 $\mu\text{g g}^{-1}$). Mean levels of Cu were equally high in all fish samples except F₄, F₅ and F₆. The coefficient of variation (C.V) also remained highest for Pb in F₃ (43.52%) while that for Cu in F₆ was the least (0.00%). When compared with other research findings and international standards, the results revealed that the streams investigated were relatively polluted by lead, copper and iron.

INTRODUCTION

Heavy metals are common components of natural waters and although some are essential for living organisms, others may become highly toxic when present in high concentration (1). When present in natural waters, these metals could enter the food chain, become bioaccumulated in aquatic organisms including fish and later become biotransferred to man, the ultimate consumer of these organisms. Moreover, metals cannot be degraded biochemically in nature. The stability of these metals therefore allows them to be transported for considerable distances by water. As a result of this process, the levels of metals in the upper members of a food

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chain can reach values sufficiently high to become health hazards when such organisms are used as food.

The pollution of our natural waters by metals commonly results from the discharge of industrial effluents into these waters (2). Such effluents often accelerate the degradation of the water quality and increases the mortality rate of aquatic organisms living in the waters.

The uptake of metals in fish and fish organs and the potential use of fish in environmental assessment programmes depend on factors such as species of fish, fish organs, modes of metal uptake and release and environmental factors (3). Metal concentrations in fish have also been related to morphology of fish, differences among species and fish weight within populations (4-7). Thus acting singly or in contact with other factors, pollutants such as toxic metals could cause large numbers of organisms to die more or less simultaneously. Hester (8) identified some toxic metals including Pb and Cu to be potentially hazardous to aquatic organisms or to man through consumption of such organisms. Many recent works on water pollution have shown that heavy metals are the commonest contaminants and that these metals often cause damage to aquatic plants and animals particularly fish (1 and 2). For a country like Nigeria where fish is the commonest source of protein, contamination of aquatic environments deserve serious attention. All the streams found within Uyo ravine (which receive most of the municipal run off and effluents discharge from the University of Uyo and a Public Hospital at Anua) finally empty into Ntak Inyang river which is a tributary of Cross River that flows into the Atlantic Ocean. Since fish is a reliable bioindicator of pollution, the determination of heavy metal levels in fish from the ravine could serve as a reliable method of assessing the extent of contribution to the pollution load of the Ntak Inyang river and hence of Cross River by the already polluted streams within the ravine.

Primarily, the aim of this study was to determine for nutritional purposes the content of Pb, Cu and Fe in seven fresh water fish commonly found in some streams within a ravine in Uyo municipality. Attempt was not made at correlating toxic metals and degree of contamination of the streams. Moreover the control of contamination of these streams is very difficult since the streams do not belong to a particular interest group or community.

MATERIALS AND METHOD

Several species of fish were randomly sampled over a period of six months from four different streams within the Uyo ravine (Fig. 1). Three sampling gears including the use of hook and line, nets and local traps (Ikpa) were employed. The nets and local traps (Ikpa) were usually set at dusk and inspected at dawn (9).

After collection, the fishes were usually put into a portable aquarium and taken to the laboratory for identification and classification and later stored in a refrigerator prior to metal determination (10). A total of seventy specimens representing seven species were selected and used for the investigation.

Each fish was filleted both sides using a stainless knife. The muscle was homogenized by grinding it in a mortar. The homogenized sample (5.0g) was digested with a mixture of trioxonitrate (V) acid (HNO_3) and tetraoxochlorate (VII) acid (HClO_4) in the ratio of 2: 1 (i.e. 10cm^3 of HNO_3 to 5cm^3 of HClO_4). The mixture was then evaporated in a sand bath inside a fume cupboard. The white cake residue formed was dissolved in 10cm^3 of nitric acid (20%). The solution was then finally diluted to 30cm^3 with ultrapure water and later analysed using an atomic absorption spectrometer (A.A.S), Unicam 9:9. Solar system. No correction for background absorption was made since the background absorption was found to be very low. Detection limits were: Pb (0.01), Cu (0.004) and Fe (0.006). Recoveries of over 98% were obtained for each element.

RESULTS AND DISCUSSION

Table 1 contains the mean concentrations and range as well as the standard deviations and coefficients of variation of the three metals (Pb, Cu and Fe) in each of the seven species ($F_1 - F_7$) of fish collected within the sampling site (Fig. 1). Fig. 2 gives the comparison of mean metal concentrations in the fish samples. Fig. 3 illustrates the coefficients of variation of metals in the fish samples.

There seems to be no particular concentration pattern in the different fish species sampled. In *Malapterurus electricus* (F_1), mean Pb level was the highest ($13.12\mu\text{gg}^{-1}$) while mean Cu content was the least ($1.07\mu\text{gg}^{-1}$). Lead (Pb) also had the highest coefficient of variation of 18.83% (Table 11). In *Auchenogonis fasciatus* (F_1), Pb content was still the highest ($6.56\mu\text{gg}^{-1}$) while Cu was the least ($0.45\mu\text{gg}^{-1}$). The coefficient

of variation was equally the highest for Pb (16.31%) followed by that for Fe (13.29%). In the case of *Brienomyrus brachyistius* (F₃), Fe had the highest mean value of $3.31\mu\text{g g}^{-1}$ followed by Pb with a mean level of $2.16\mu\text{g g}^{-1}$. However, the coefficient of variation remained the highest for Pb (43.52%) followed by Fe (11.78%). *Chromidotilapia guntheri* (F₄) recorded the highest mean Fe content of $3.31\mu\text{g g}^{-1}$ followed by a mean concentration of $4.53\mu\text{g g}^{-1}$ for Pb and $0.10\mu\text{g g}^{-1}$ for Cu. Surprisingly, Cu had the highest coefficient of variation of 20.00%. In *Hepsetus odoe* (F₅), Fe content was the highest ($11.25\mu\text{g g}^{-1}$), followed by Pb ($6.90\mu\text{g g}^{-1}$). However, the coefficient of variation for Pb was the highest (10.00%), followed by that for Fe (3.38%). In the cases of *Papyrocranus ager* (F₆) and *Bryanus longipinnus* (F₇), Pb levels were the highest ($12.04\mu\text{g g}^{-1}$) and ($8.76\mu\text{g g}^{-1}$) respectively while Cu contents were the least ($0.00\mu\text{g g}^{-1}$) and ($0.07\mu\text{g g}^{-1}$) respectively (Table 1). The coefficients of variation was found to be comparatively much higher (42.86%) for Cu in *Brycinus longipinnus* (F₇) species than in *Papyrocranus ager* (F₆) species (0.00%).

Considering the metal levels in the species, Pb was found to exhibit the following trend: $F_1 > F_6 > F_7 > F_5 > F_2 > F_3$ (Table 1). For Cu, the trend was $F_5 > F_1 > F_2 = F_3 > F_4 > F_7 > F_6$ while for Fe, the trend was $F_5 > F_1 > F_7 > F_6 > F_2 = F_3 = F_4$ (Table 1). There was therefore no definite or regular trend in the metal levels in all the fish species analysed.

Since the Pb concentration in all the fish species fell outside the range of $0.003 - 0.07\mu\text{g g}^{-1}$ obtained for fish from unpolluted waters in other parts of the world, the streams within Uyo ravine could be said to be polluted by Pb (11 - 13). Moreover, based on standards set up by the Australian National Health and Medical Research Council (NMRC) for Pb ($2.0\mu\text{g g}^{-1}$) in sea food, the mean levels of $13.12\mu\text{g g}^{-1}$ in F₁, $6.56\mu\text{g g}^{-1}$ in F₂, $2.16\mu\text{g g}^{-1}$ in F₃, $4.53\mu\text{g g}^{-1}$ in F₄, $6.90\mu\text{g g}^{-1}$ in F₅, $12.04\mu\text{g g}^{-1}$ in F₆ and $8.76\mu\text{g g}^{-1}$ in F₇, samples are high enough to cause serious health problems to individuals who may happen to consume them (2 and 14).

This deduction is arrived at following the experimental evidence for the existence of carcinogenicity (renal tumours) of inorganic lead in rat after oral ingestion of high doses of Pb (15). Moreover, at concentration lower than 0.05 ppm which is the maximum total lead permitted in drinking water in the United States and Nigeria, fish may suffer toxic effects or become adversely affected (16 and 17).

The levels of Cu in all the fish samples particularly F₁, F₂, and F₃, were quite high and worrisome. Cu is almost always elevated in the gills,

Liver, pancreas and other internal organs (15). Copper is one of the most toxic heavy metals to fish and its toxicity depends on the degree of hardness of the water in which they live. The LC₅₀ of copper typically ranges between 0.02 and 1.0mg Cu/L in fresh water (18). An increase in water hardness to approximately 500mg/L as CaCO₃ is said to increase the enhanced synergistically by combinations of C/Al/Zn, Cu/Cd/Zn, Cu/Zn, Cu/Zn/Ni, Cu/Zn/phenol, and Cu/H⁺ (18). However organic sequestration always significantly reduces acute toxicity. Therefore although several species of fish have the ability to adapt to potentially toxic copper levels, at least temporarily, yet this ability reduces if the water is reasonably hard owing to the presence of CaCO₃, ionic (Cu²⁺) and its hydroxides (Cu₂OH₂²⁺, CuOH⁺) which are very toxic (18) There are however only a few cases of acute poisoning by Cu and its salt. The chief symptoms following ingestion of Cu are epigastric burning, nausea, vomiting, and diarrhea (19). There may also be lesions in the gastrointestinal tract and induction to hemolytic anemia. On the whole, chronic Cu poisonings is rarely reported, except for individuals suffering from a congenital disorder known as Wilson's disease.

The levels of Fe were generally quite high to warrant concern particularly in F₁, F₂, F₃, and F₇ species. However, Fe is rarely a toxicologically significant contaminant of fish tissue and often occur in high concentration (20-22). Since the lethal concentration (LC₅₀) of total Fe to fish generally range from 0.3 to >10mg/l, depending on species and test conditions, the levels of Fe in the fish samples could lead to some of the health effects such as vomiting, gastrointestinal bleeding, pneumonitis, convulsion, coma and jaundice often associated with acute exposure to Fe (15). In some cases, a generalized increase in Fe content (known as Hemosiderosis) occurs, in other instances, a specific deposition may lead to localized fibrosis. Although this condition is relatively benign, some studies have indicated that it is accompanied by abnormal glucose metabolism and increase heart disease.

CONCLUSION

The results of the research have revealed that the streams within the ravine are to some extent polluted by the heavy metals Pb, Cu and Fe. It is suspected that the metals may have been introduced into these streams by the discharges into the ravine by small scale industries, production.

industries, other industries associated with these metals and run off through refuse dumps, mechanic workshops, petrol filling stations and agricultural farms. Infact Udosen et al (23) had earlier warned of the dumping of paint wastes into this ravine. The metals so accumulated by the fish in these streams could become biomagnified in the muscles and later biotransferred to consumers. The result could be undue interference with the normal physiological functioning of the body which in turn could cause untold damages ranging from nervous malfunctioning to death.

TABLE 1. METAL CONCENTRATIONS ($\mu\text{g g}^{-1}$) IN FISH SAMPLES

S/N	SAMPLE	METALS/CONCENTRATIONS			
		Pb	Cu	Fe	
1	F ₁	Mean	13.12	1.07	11.03
		Range	9.05-16.11	0.98-1.20	10.01-11.90
		Std. Dev.	2.47	0.07	0.47
		Coeff. Of Variation	18.83%	5.54%	4.26%
2	F ₂	Mean	6.56	2.45	8.31
		Range	5.00-8.30	0.60-3.50	2.77-4.00
		Std. Dev.	1.07	0.03	0.44
		Coeff. Of Variation	16.31%	6.67%	13.29%
3	F ₃	Mean	2.16	0.45	3.31
		Range	1.03-3.60	0.40-0.50	2.78-3.79
		Std. Dev.	0.94	0.03	0.39
		Coeff. Of Variation	43.52%	6.67%	11.78%
4	F ₄	Mean	4.53	0.01	3.31
		Range	3.60-5.60	0.07-0.15	2.79-86
		Std. Dev.	0.73	0.02	0.36
		Coeff. Of Variation	16.11%	20.00%	10.88%
5	F ₅	Mean	6.90	1.16	11.25
		Range	5.88-8.00	1.13-1.19	10.67-11.90
		Std. Dev.	0.69	0.02	0.38
		Coeff. Of Variation	10.00%	1.72%	3.38%
6	F ₆	Mean	12.04	0.00	5.07
		Range	10.33-13.05	0.00	4.50-5.90
		Std. Dev.	0.91	0.00	0.36
		Coeff. Of Variation	7.56%	0.00%	6.90%
		Mean	8.76	0.07	9.71

7	F ₇	Range	7.50-9.98	0.04-0.12	9.05-11.30
		Std. Dev.	0.65	0.3	0.63
		Coeff. Of Variation	7.48%	42.86%	6.49%

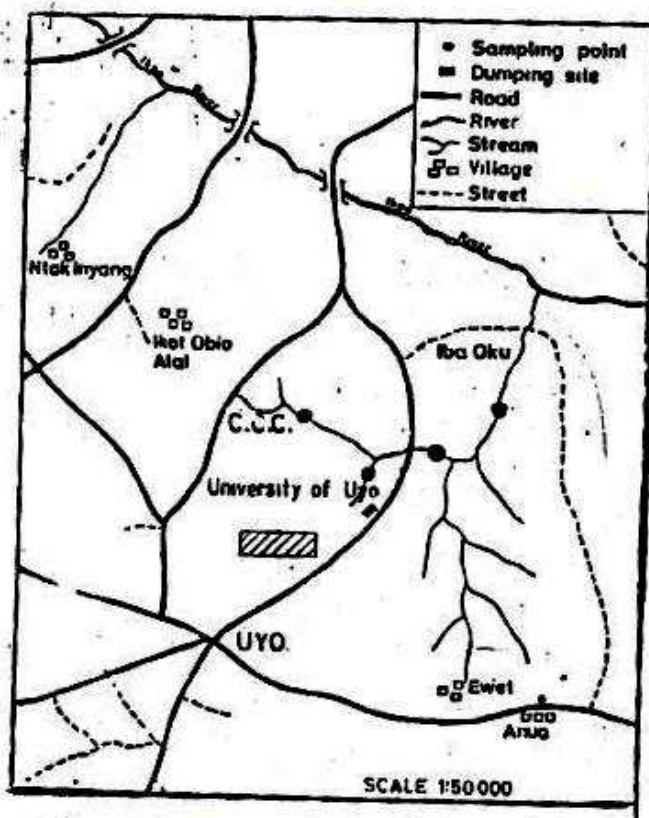


Fig. 1. Part of Uyo showing the sampling site.

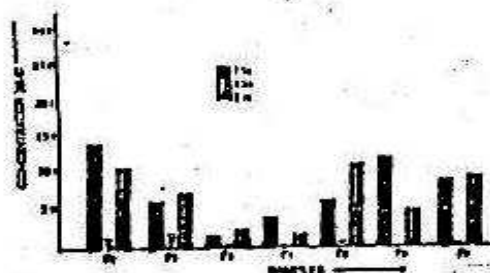


Fig. 2. Comparative levels of metals in the fish samples.

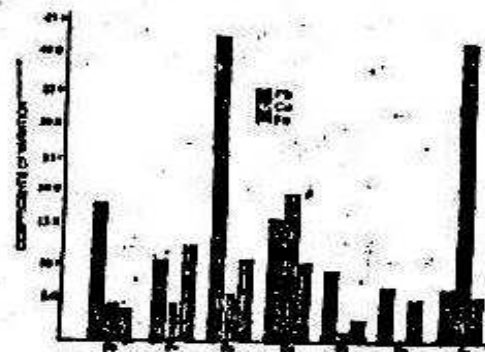


Fig. 3. Comparative coefficients of variation of metals in the samples.

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